

## Quiz 2: Process intensification

### Part 1: True-false

**Circle T (true) or F (False)**

- 1)        **T    F**        For single reactant first order reactions of type  $A \rightarrow B$ , there is no influence of the degree of segregation on the conversion of reactant A.
- 2)        **T    F**        For a same concentration set using the Villermoux-Dushman protocol, a decrease in the segregation index implies an increase in the mixing time.
- 3)        **T    F**        For a homogeneous reaction, the characteristic reaction time is linearly dependent on the characteristic dimension of the reactor.
- 4)        **T    F**        The Nusselt number is significantly higher in a COBR than in an unbaffled tube (under identical  $Re_n$  and  $Re_o$  conditions).
- 5)        **T    F**        In the laminar regime, the Bodenstein number in a COBR is higher than in a smooth (unbaffled) tube (using identical  $Re_n$  and  $Re_o$ )

Short justification of your answer:

---

---

---

- 6)        **T    F**        In a RPB, the HETP is significantly higher than in a conventional packed column.
- 7)        **T    F**        In a RPB, the liquid is delivered at the eye of the rotor.
- 8)        **T    F**        A RPB can be used for stripping, gas absorption and distillation.
- 9)        **T    F**        In a TF-SDR, the film thickness decreases with flowrate.
- 10)       **T    F**        In a TF-SDR, the film thickness increases with disk rotational speed.
- 11)       **T    F**        In a TF-SDR, the mixing time is short enough to carry out precipitations and obtain very low particle sizes with particularly narrow particle size distributions.
- 12)       **T    F**        In a TF-SDR, the film thickness decreases with radial position.
- 13)       **T    F**        In a PFR, there is no effect of segregation on the conversion of single reactant reactions of the type  $A \rightarrow P$ .

- 14)                      T      F              For a single reactant reaction of the type  $A \rightarrow P$  with a reaction order  $n > 1$ , a negative effect of segregation on the reaction rate is expected.

### Part 2: Multiple choice

Choose the correct answer. Check only one box per question, as there is only one correct answer.

- 
- 15) The ratio of characteristic times between a homogeneous and a heterogeneous reaction ( $R$  = characteristic length of the reactor)
- ☐ is independent of  $R$                       ☐ increases with  $R$                       ☐ decreases with  $R$
- 
- 16) For the following competing scheme:  $A_1 + 2A_2 \rightarrow A_3$  (instantaneous) and  $A_4 + 3A_2 \rightarrow A_5$ , with  $c_{10} = 1$  and  $c_{40} = 2$ ,  $Y_{CS}$  is equal to
- ☐ 3/5                      ☐ 3/4                      ☐ 2/3
- 
- 17) At equal values of  $t_{mx}$  and  $\tau$ , the segregation intensity in a CSTR is
- ☐ larger than in a PFR                      ☐ smaller than in a PFR                      ☐ the same as in a PFR
- 
- 18) For the following reaction:  $A_1 + A_2 \rightarrow P$ , first order in  $A_1$  and  $A_2$  with  $DaI_{mx} = 1$  (separate  $A_1$  and  $A_2$  feeds), the conversion in a given type of reactor at a given value of  $DaI$ , compared to the micromixed system, is expected to be
- ☐ higher                      ☐ lower                      ☐ the same
- 
- 19) For a single reactant reaction of the type  $A \rightarrow P$  carried out in a PFR, segregation has the following effect on conversion (compared to a micromixed PFR):
- ☐ Increases the conversion                      ☐ Decreases the conversion                      ☐ No effect on conversion                      ☐ Depends on the reaction order
- 
- 20) In a PFR, the intensity of segregation
- ☐ increases with  $Z$                       ☐ decreases with  $Z$                       ☐ is constant in the reactor
- 
- 21) The segregation index was measured in two reactors using the Villermoux-Dushman protocol.  $X_s(\text{reactor } 1) > X_s(\text{reactor } 2)$ . Which reactor has the lowest mixing time?
- ☐ Reactor 1                      ☐ Reactor 2                      ☐ There is not enough information available to decide

Short justification of your answer:

---

22) Which variables should preferably be kept constant for the scale-up of a COBR?

☐  $\frac{L}{D}, \alpha, \psi, Str$ ☐  $\frac{L}{D}, Re_o, Str, \psi$ ☐  $\frac{L}{D}, \alpha, f, x_o$ 

---

23) The RTD in a RS-SDR can be described by the following model:

☐ Plug-flow☐ Combination of plug-flow and cascade of CSTRs☐ Cascade of CSTRs

---

24) For a reaction with  $\Delta V^\ddagger > 0$ , an increase in the pressure

☐ increases the reaction rate☐ decreases the reaction rate

---

25) How do the following properties change when a liquid changes from standard (below critical  $p$  and  $T$ ) to supercritical conditions?

 $D_m$ ☐ Increases☐ Decreases $\rho$ ☐ Increases☐ Decreases $\mu$ ☐ Increases☐ Decreases

### Solutions

1T, 2F, 3F, 4T, 5T (creation of radial velocity components in the COBR chambers which flatten the parabolic velocity profile; this is due to vortices forming when baffles interact with the flow), 6F, 7T, 8T, 9F, 10F, 11T, 12T, 13T, 14F.

15C, 16B, 17A, 18B, 19C, 20B, 21C (Segregation index can only be compared at equal concentration sets, which are not specified here), 22A, 23B, 24B, 25ABB